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## Elements of Chemical Reaction Engineering

So that it is easier to visualize the solution, we first plot the inverse of the reaction rate versus conversion. This type of plot is often called a "Levenspiel Plot.": Recalling the mole balance equations for a CSTR and a PFR: Until the conversion (X) reaches 0.5, the reaction rate is independent of conversion and the reactor volumes will be ...

## Elements of Chemical Reaction Engineering

If a reaction rate doubles for an increase in 10°C, at  $T = T_1$  let  $k = k_1$  and at  $T = T_2 = T_1 + 10$ , let  $k = k_2 = 2k_1$ . Then with  $k = Ae^{-E/RT}$  in general,  $k_1 = Ae^{-E/RT_1}$  and  $k_2 = Ae^{-E/RT_2}$ , or

## Solutions for Chapter 3 – Rate Law and Stoichiometry

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