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4.8.The rate of the chemical reaction doubles for and increase of 10 K in absolute temperature from 298 K. Calculate E a. Ans. 4.9.The activation energy for the reaction, 2 Hl(g) —> H 2 +l 2 (g) is 209.5 k J mol -1 at 581 K.Calculate the fraction of molecules of reactants having energy equal to or greater than activation energy?

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