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Limiting Reagent Made Easy: Stoichiometry Tutorial Part 5
Most Common Chemistry Final Exam Question: Limiting Reactants Review *Chemistry Practice Problems: Limiting Reactants* **Practice Exercise p 101 Limiting Reactant Calculations with Moles**
Theoretical, Actual, Percent Yield \u0026amp; Error - Limiting Reagent and Excess Reactant That Remains

Skill Practice 33 Limiting Reactants

The limiting reactant is the amount of a chemical substance that is consumed, as well as limits how much product forms, during a chemical reaction. In chemistry, calculating the limiting reactant...

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In this case, oxygen is called the “limiting reactant” because it limits how much propane can react. Notice that the limiting reactant isn’t always the substance that is present in the fewest number of moles. In this example, propane (C₃H₈) is the “excess reactant” because after the reaction there will be some of it left over.

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