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 ϵ_0 , where $\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{N}\cdot\text{m}^2$, and $a = F/m$
eE/m, where Physics 42 HW Set 1 Serway
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Solutions 23.1 (a) $N = 10.0 \text{ grams} / 107.87 \text{ grams mol}^{-1} \times 6.02 \times 10^{23} \text{ atoms mol}^{-1} = 47.0$
electrons atom = $47.0 \times 2.62 \times 10^{24}$ (b) # electrons

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added = $Q_e = 1.00 \times 10^{23} \text{ C}$ $1.60 \times 10^{-19} \text{ C}$
electron = 6.25×10^{15} or 2.38 electrons for
every 10^9 already present 23.2 (a) F_e

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107.87 grams mol 6.02×10^{23} atoms mol 47.0
electrons atom = $\times 2.62 \cdot 10^{24}$ (b) # electrons
added = $Q / e = 1.00 \times 10^{23} \text{ C} / 1.60 \times 10^{-19} \text{ C}$
electron = 6.25×10^{15} or 2.38 electrons for
every 10^9 already present 23.2 (a) $F_e = k_e \frac{q_1 q_2}{r^2} = 8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2 \frac{1.60 \times 10^{-19} \text{ C}}{2} (3.80 \times 10^{-10} \text{ m})^2 = 1.59 \times 10^{-9} \text{ N}$ (repulsion)
(b) $F_g = G \frac{m_1 m_2}{r^2}$

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7cq; 7oq; 7p; 8cq; 8oq; 8p; 9cq; 9oq; 9p;
10cq; 10oq; 10p; 11cq; 11oq; 11p; 12oq; 12p;
13oq; 13p; 14oq; 14p; 15oq; 15p; 16p; 17p;
18p; 19p; 20p; 21p; 22p; 23p; 24p; 25p; 26p;
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