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except – oxides,
carbonates,

bicarbonates and
metallic carbides.

Plants, Animals,
Petroleum, dyes and
drugs are all natural
sources.

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Covalent Bond 1. The hybridization of the central carbon in $\text{CH}_3\text{C}\equiv\text{N}$ and the bond angle CCN are
a. sp^2 , 180° . b. sp , 180° . c. sp^2 , 120° . d. sp^3 , 109° . 2. Which of the following statements about an sp hybridized carbon

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is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

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What is the hybridization at nitrogen? Question2:
How many

stereoisomers of phenylpropylene oxide are there?

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- 1)Cyclotetrane
- 2)Benzene
- 3)Naphthalene
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Solution: Dipole
moment of CCl_4 is 0.
The order of dipole
moment is $\text{CCl}_4 <$
 $\text{CH}_3\text{Cl} < \text{CH}_2\text{Cl}_2 <$
 CHCl_3 . Hence option
(3) is the answer. 4.

Which one of the

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following compounds
will not be soluble in
sodium bicarbonate ?

- (1) Benzene
- sulphonic acid (2)
- Benzoic acid (3) o-
- Nitrophenol (4) 2, 4, 6
- Trinitrophenol.

Solution:

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compound is given
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methane, ethyne, and
ethene. From the list
above, name a
compound : (i)
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