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Scientific Notation 1. Express each of the following numbers in scientific notation. a. 230 b. 5901 Use with Appendix B, Scientific Notation 100000 d. 56 million e. $\frac{2}{10}$ f. 0.4 0 13 g. 0.0 h. 0.000 i. 0.000000 8 XID OOD 6.0 j. 12 thousandth 1, I KID 2.

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Answers MATH HANDBOOK TRANSPARENCY MASTER

Operations with Scientific Notation Addition and Subtraction

Before numbers in scientific notation can be added or subtracted, the exponents must be equal. Not equal Equal $(3.4 \times 10^2) + (4.57 \times 10^3) = (0.34 \times 10^3) + (4.57 \times 10^3)$ The decimal is moved to the left to increase the exponent.

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Mcgraw Hill Chemistry Worksheets - Teacher Worksheets

MATH HANDBOOK TRANSPARENCY WORKSHEET

Operations with Scientific Notation I. Perform the following operations and express the answers in scientific notation.

a. $(1.2 \times 10^5) + (5.35 \times 10^6)$ S S
b. $(6.91 \times 10^{-2}) + (2.4 \times 10^{-3})$
c. $(9.70 \times 10^6) + (8.3 \times 10^5)$
d. $(3.67 \times 10^2) - (1.6 \times 10^1)$
e. $(8.41 \times 10^{-5}) - (7.9 \times 10^{-6})$
f. $(1.33 \times 10^5) - (4.9 \times 10^4)$ 2.

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Scientific notation takes the form of $M \times 10^n$ where $1 < M < 10$ and "n" represents the number of decimal places to be moved. Positive n indicates the standard form is a large number. Negative n indicates a number between zero and one. Example 1: Convert 1,500,000 to

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scientific notation,

SCIENTIFIC NOTATION Name Example 1: Convert 1,500,000 to

...

Math Handbook Transparency Masters Chemistry: Matter and
Change 93 Addition and Subtraction Before numbers in scientific
notation can be added or subtracted, the exponents must be equal.
(3.4×10^2) + (4.57×10^3) ≠ (0.34×10^3) + (4.57×10^3) Not equal
Equal The decimal is moved to the left to increase the exponent. #
(0.34×4.57) $\times 10^3$ ≠ 4.91×10^3 Multiplication

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worksheets found for this concept. Some of the worksheets for this concept are Operations scientific notation, Scientific notation mixed operations es1, Writing scientific notation, Writing numbers in scientific notation, Bastien, Name date class 53 math handbook transparency master 1, Scientific notation muldiv es1, Operations in scientific notation.

Operations With Scientific Notation Answers Worksheets ...

94 Chemistry: Matter and Change Math Handbook Transparency Worksheets 1. Perform the following operations and express the answers in scientific notation. a. $(1.2 \times 10^5) \div (5.35 \times 10^{-1})$ $(1.2 \times 10^5) \times (5.35 \times 10^6)$ $(0.12 \times 10^6) \div (5.35 \times 10^6)$ $(0.12 \times 5.35) \times 10^6$ 5.47×10^6 b. $(6.91 \times 10^2) \div (2.4 \times 10^3)$ $(6.91 \times 10^2) \times (2.4 \times 10^3)$ $(6.91 \times 10^2) \div (0.24 \times 10^2)$

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Scientific Notation Use with Appendix B, Scientific Notation

96 Chemistry: Matter and Change Math Handbook Transparency Worksheets 1. For each of the measurements in the table below, determine if the underlined number is significant or not significant. Place a check mark in the appropriate box and in the box under the rule that you used to make your determination. 2.

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