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electric attraction between nuclei and valence electrons of
different atoms that binds the atoms together. electric
attraction. positive charges are attracted to negative charges.

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covalent bonds . electron pairs are shared between 2 atoms, normally nonmetals. polar-covalent bond ...

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that results from the electrical attraction b ...

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CHAPTER 6 REVIEW Chemical Bonding SECTION 1

SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

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Answers

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The Chemical Bonding chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with chemical bonding.

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