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Thermodynamics, $3e \quad N \quad V \quad R T \times 18 \text{ mol} \times 8.314 \times 10^{-5} \text{ (bar m}^3 \text{ mol}^{-1} \text{ K}^{-1})$
 $T P = \frac{V V_1 \times 10^{-3} - (10 - x) 10^6}{V}$ volume taken up by liquid Equating P and P_{vap}
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Chapter 1 Solutions. Engineering and Chemical Thermodynamics. Wyatt Tenhaeff Milo Koretsky. Department of Chemical Engineering Oregon State University. koretsm@enr.orst.edu

1.2 An approximate solution can be found if we combine Equations 1.4 and 1.5:

$$\frac{1}{2} m v_{\text{ekmolecular}}^2 = \frac{3}{2} kT_{\text{ekmolecular}}$$

2. $3kT = V m$. Assume the temperature is 22 C.

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