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$1/2$ . (b) Let  $h$  be the half-life, that is, the amount of time it takes for a quantity to decay to one-half of its original amount. Since  $r_0 = 1/2$

$$1/2 = r_0 e^{-h/2} \Rightarrow r_0 = r_0 e^{-h/2}$$

$$r_0 = r_0 e^{-h/2} \Rightarrow 1/2 = e^{-h/2}$$

$$-\ln 2 = -h/2 \Rightarrow h = 2 \ln 2$$

In Exercise 7, we stated that the half-life of Carbon 14 is 5230 ...

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