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~~28. Crystal field theoryCoordination number and Oxidation state Coordination Chemistry | Monday MCQ | Detailed Solution Crystal Field Theory Crystal Field Theory (Octahedral Geometry) for Coordination Compounds Coordination Compounds || Valence Bond Theory || Class-12 || C.B.S.E~~ Lecture: Coordination Compounds - 1 **IUPAC name of coordination compound Molecular Orbital Approach to Coordination complexes IUPAC Nomenclature of Coordination Compounds Class 12 | Narendra Sir (IITB 2003, Purdue Univ USA) Valence Bond Theory (VBT) || Coordination compounds | octahedral, tetrahedral, square planar complex**

~~Chemistry 107. Inorganic Chemistry. Lecture 23.~~

~~(L-9) CFT (crystal field theory) + Limitation of VBT | Coordination Compounds | By Arvind AroraControl and Coordination Class 10 Science Chapter 7 - Explanation, Question Answers Accelerate NEET 2020 | Super Tricks To Remember p-Block Reactions | Chemistry | Ashwani Sir |Gradeup PYQs on Coordination Chemistry | Chemical Sciences | Unacademy Live - CSIR UGC NET | Jagriti Sharma Coordinationchemistryii Ppt~~

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~~Common Ligands. Monodentate ligands donate one pair of electrons to the central metal atoms. An example of these ligands are the halide ions (F-, Cl-, Br-, I-).Polydentate ligands, also called chelates or chelating agents, donate more than one pair of electrons to the metal atom forming a stronger bond and a more stable complex.~~

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