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Phase Change

Lab, Heating and

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for a Liquid-

Solid System

Cooling Curve of

Naphthalene

Reading Heating

and Cooling

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~~Chemistry~~

~~Cooling curves~~

~~stearic acid~~

~~cooling from a~~

~~liquid to a~~

~~solid Cooling~~

~~Curve t-Butanol~~

~~Lab 5 1 3 Lab~~

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stars Observing
the cooling of
stearic acid

~~Lauric Acid~~
~~Cooling Curve~~
~~2.5~~

~~Heating/Cooling~~
~~Curves~~

~~(Potential and~~
~~Kinetic Energy~~
~~Changes) Phase~~
~~Change and~~
~~Heating Cooling~~
~~Curve Problem~~

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(Chemistry
Lecture)
~~Answers~~
~~Activity C3~~

Demonstration of
how to Heat and
cool steric acid

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: Properties of

Pure Substances

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Results for the
cooling curve of
paraffin wax:

Graph showing
the cooling
curve of

paraffin wax: As
the time
increases, the
temperature of
the paraffin wax
decreases.

Therefore, in

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Lab Chemistry, the
Answers
paraffin wax in
its liquid state
changes to solid
state as more
time goes by due
to the decrease
in temperature.

*Unit 2
assignment 2 -
Lab Report on
Study of Cooling
curves*

Get Free Cooling Curve Chemistry 1 Experiment #1: The Cooling Curve of Stearic Acid

INTRODUCTION

Matter around us exists in three common states—solid, liquid, and gas. Matter can change from one state (or phase, as it is

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sometimes called) to another. Ice, for example, is the solid state of H₂O. Add

*Chemistry 1
Experiment #1:
The Cooling
Curve of Stearic*

...

Materials: test
tube,

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naphthalene,
thermometer, hot
water, burner,
stand, clamp. Be
careful as
naphthalene
vapour is
poisonous: you
can wear a
breathing mask
to reduce the
risk of you
inhaling the
vapour Method.

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Fill a test tube with naphthalene until it is about a quarter full; Place the test tube in hot water to allow the naphthalene to melt

*Experiment: the cooling curve of naphthalene -
Free ZIMSEC ...*

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Answers Heating
and Cooling

Curves - AP

Chemistry

CHEMISTRY 85 80

75 70 65 60 55

Temp. (QC) 5 0

40 35 30 25 20

15 10 10 -15 -20

Date Class

HEATING CURVE

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WORKSHEET

Heating Curve of
Substance X

12 14 16 18 20

22 24 26 28 30

Time (Minutes)

The heating
curve shown
above is a plot
of temperature
vs time.

Cooling Curve

Lab Chemistry

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A cooling curve for salol The temperature stays the same during a state change: during melting and evaporating, internal energy increases as the motion of particles increases and

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bonds are
broken...

*Cooling curves -
Physical changes
- KS3 Physics
Revision ...*

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up in harmful
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reading a good

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of coffee in the
afternoon,

instead they are
facing with some
harmful virus

inside their
laptop. Cooling
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Chemistry

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instantly.

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play this quiz,
please finish
editing it. 8

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Questions Show
answers.
Question 1

Chemistry
Heating &
Cooling Curves
WCHS Quiz -
Quizizz
Cooling curves
are the
opposite. They
show how the
temperature

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changes as a
substance is
cooled down.

Just like
heating curves,
cooling curves
have horizontal
flat parts where
the state
changes from gas
to liquid, or
from liquid to
solid. You are
likely to have

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stearic acid in
a school

practical lesson
to make your own
cooling curve.

*Heating and
Cooling Curves -
Mr. Kent's
Chemistry
Regents ...*

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Change Lab,
Heating and
Cooling Curves
Heating and
Cooling curves
for the
freezing/melting
of a pure
substance Lab 2a
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A heating curve
of a substance
gives the
changes in
temperature as

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Lab Chemistry
Answers

we move from a solid to a liquid to a gas. A cooling curve gives the changes in temperature as we move from gas to liquid to solid.

*Formal
experiment 1:
Heating and*

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*cooling curve of
water ...*

The cooling
curve, a plot of
temperature
versus cooling
time, in Figure
\(\PageIndex{4}\
) plots
temperature
versus time as a
75 g sample of
steam, initially
at 1 atm and

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200°C, is cooled. Although we might expect the cooling curve to be the mirror image of the heating curve in Figure $\left(\backslash\text{PageIndex}\{3\}\backslash\right)$, the cooling curve is not an identical mirror image. As heat is removed from

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the steam, the temperature falls until it reaches 100°C .

*11.7: Heating
Curve for Water
- Chemistry
LibreTexts*

For my chemistry lab, I heated stearic acid at a constant rate until it

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Lab Chemistry into
Answers
a liquid and
then reversed
the process by
cooling to
produce a solid
again. In this
experiment, 1)
What is going on
at the molecular
level as the
liquid stearic
acid cools and
freezes?? 2) How

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would an increase in the amount of stearic acid used affect the shape of the heating and cooling curves?

*Stearic Acid
Heating &
Cooling Curves?
| Yahoo Answers*
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Cooling Curve of
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-Curve-Worksheet-
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at Valdosta High
School. STUDENT
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cool? A beaker
of hot water is
put into a bowl
of cold water.

The hot water is
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eet-Answer-
Key.doc ...

*Heating And
Cooling Curves
Answers |
voucherslug.co*

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– Yes – change
in potential
energy. – Phase
changes are
going on, and as
it is melting,
it indicates
that there is a
change in
potential energy
– known as there
is a change in
position of
atoms. – The

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temperature
remains the same
at the plateau;
therefore, there
is no change in
kinetic energy.
Click again to
see term ?

*Chemistry H, Icy
Hot Lab! :)
Flashcards /
Quizlet
Heating Curve*

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For Water Lab ...
Getting to the
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(The Basics)
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cooling curve
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Lab Chemistry

Activity 1-At a
Glance CHEMISTRY

HEATING CURVE

WORKSHEET

Plotting a

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Graphing A

Heating Curve

For Water Lab

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