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**PART 14 COLLISION THEORY**

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Based on your answer to Key Question 5, complete the following statement:

Collision theory states that a reaction is

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most likely to occur if&mldr; If the reactants collide with the correct amount of energy and correct orientation. Exercise 1. Hydrogen gas and iodine vapor combine to form hydrogen iodide gas, as shown in the equation  $H_2 + I_2 \rightarrow 2 HI$ .

**Collision\_Theory\_pogil\_questions -**

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## **Collision Theory Key ...**

Download File PDF Collision Theory Pogil Solution Collision Theory Pogil Solution Based on your answer to Key Question 5, complete the following statement: Collision theory states that a reaction is most likely to occur if&mldr; If the reactants collide with the correct

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amount of energy and correct orientation.  
Exercise 1. Hydrogen

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collision theory is that, in order to react, molecules must collide. This fundamental rule guides any analysis of an ordinary

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reaction mechanism. Collision Theory Pogil Solution The collision theory states that a chemical reaction can only occur between particles when they collide (hit each other). The collision between reactant particles is necessary but not

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Collision Theory Model: Collision Theory

In the picture below, the baseball bat represents Reactant A and the baseball represents Reactant B. A reaction will only be successful if the batter hits a homerun.

If the batter does not hit a homerun, the reaction will be considered a failure. Now,

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particles is necessary but not sufficient for a reaction to take place. The collisions also have to be effective.

## **Collision Theory - Impact for a Chemical Reaction**

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necessary but not sufficient for a reaction to take place.

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