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CHEM 131 L- Gas Laws

- The questions and
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CHEM 131 Lab- Blue

Dye - The questions and
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CHEM 131- chemical reactions
CHEM 131- exp 5 Synthesis of
Limiting Reactant Chem
131- Post Lab 1 - The
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EXP 13 - The questions
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~~CHEM 131 L- Experiment
9- The questions and
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Answers to Post-Lab

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Questions for

Experiment 5. I have 40 L of methane gas measured at standard temperature and pressure (STP). How many moles of methane gas is this? 1 mole of any gas at STP = 22.4 L Moles of methane = $(40 / 22.4)$ = 1.79 moles. What is the volume of 0.30 moles of nitrogen gas if measured at STP? 1 mole of any gas at STP = 22.4 L Volume

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of nitrogen = $0.3 \times 22.4 =$
6.72 L.

~~SCC1123 Experiment 5
Post-Lab Questions
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Post lab questions

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(Chemistry of Copper)

Part 1: [Copper metal to
Copper (II) nitrate]

2pts. Balanced chemical
equation: Observations:
(type your observations
below after watching the
video titled Formation of
copper (II) nitrate. Part 2:
[Preparation of Copper
(II) hydroxide] 2pts.

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~~Copper) Part 1 ...~~

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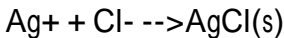
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Favorite Answer. First calculate the moles of KHP (potassium hydrogen phthalate, $\text{KHC}_8\text{H}_4\text{O}_4$). The molar mass for KHP = 204.2 g/mole. $0.531 \text{ g KHP} \times (1 \text{ mole KHP} / 204.2 \text{ g KHP}) = 0.00260$ moles KHP. KHP...

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Favorite Answer. The white precipitate most likely came from the presence of chloride ions.



The chemical responsible for the iron is Fe (iron), in some form. It could be almost...

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Follow this series of questions. When you can answer " yes: " to a

question, then stop. 1)

Does your reaction have oxygen as one of it's reactants and carbon dioxide and water as products? Yes: it's a combustion reaction 2)

Does your reaction have two (or more) chemicals combining to form one chemical? Yes: it's a

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synthesis reaction

Answers

Chemistry of Copper

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Name This assignment will guide you through an extraction procedure.

Your task is to separate a

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mixture of 3 compounds
using a series of acid-base
extractions.

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Chemistry is an
experimental science.

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Thus, it is important that students of chemistry do experiments in the laboratory to more fully understand that the theories they study in lecture and in their textbook are developed from the critical evaluation of experimental data.

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~~Manual~~

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Experiment 15

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way. in the course of
them is this experiment
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