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Dyes are normally water-soluble or water dispersible organic compounds that are capable of being absorbed into the substrate destroying the crystal structure of the substance. The dye molecules are usually

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chemically bonded to the surface and become a part of the material on which it is applied.

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Basis for colour. Unlike most organic compounds, dyes possess colour because they 1) absorb light in the visible spectrum (400–700 nm), 2) have at least one chromophore (colour-bearing group), 3) have a conjugated system, i.e. a structure with alternating double and single bonds, and 4)

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exhibit resonance of electrons, which is a stabilizing force in organic compounds (Abraham, 1977).

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4.7.3. Principles of
sulphur dye application.

The principles of
application of sulphur to
cotton are the same as
those for vat dyes. The
process consists of four
steps including
reduction of the dye to

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make it soluble,
application of the dye,
i.e. dyeing, oxidation of
the absorbed dye in the
fibre and finally rinsing
or quick washing.

Sodium sulphide is a
commonly used
reducing agent.

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Horrobin 16 has very adequately described the hydrolysis chemistry of these dyes. Figure 9.4 summarises the basic chemical principles of the DCT dye reaction with cellulose and the competing hydrolysis reactions. Download : Download full-size image; 9.4. The dye-fibre fixation

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process for a dichloro-s-triazine dye on cellulosic fibres.

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developed for polyesters

are known as disperse

dyes In this case, the

mechanism of

coloration involves

□dissolving□ the dye in

the polymer matrix to

form a solid□solid

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Therefore, these dyes are mainly used to dye cotton fibres. Indigo is an important example of this type (Table 34.1).

(viii) Mordant Dyes

These dyes require an additional substance (generally a metal ion) for fixing to the fibre.

These are used mainly for dyeing wool. The

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method involves the
precipitation of certain

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The unmasked dyes are
highly colored and
fluorescent ($\lambda = 388 \times 10$)

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4 M \times 1 cm \times 1 and fluorescence quantum yields (Φ) = 40–85 % in the unbound state and in methanol). By stepwise and orthogonal protection of carboxylic and sulfonic acid groups a highly water-soluble caged red-emitting dye with two sulfonic acid residues was prepared.

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