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States of Matter : Solid

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Pure Substances and

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atmosphere, ozone

formation,

chlorofluorocarbons,

CFC ' s 1.2 Chemistry

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Chapter 1 Section 2

MATTER and Its
Properties Matter –

Anything that has MASS
and VOLUME. ATOMS
and MOLECULES the
basic building blocks of
matter. ATOM-- the

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smallest unit of an element that still maintains the properties that element.

ELEMENT – a pure substance made up of only one type of atom.

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Section 1
Chemistry: Matter and
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Section 6.3 Periodic

Answers
Trends In your textbook,
read about atomic radius
and ionic radius. Circle
the letter of the choice
that best completes the
statement or answers the
question. 1. Atomic radii
cannot be measured ...

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CHAPTER 3

SOLUTIONS

MANUAL

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Changes Matter—Proper
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Solutions Manual

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Section 3.1 Properties of
Matter pages 70 – 75

Problem-Solving Lab 1.

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Explain why the flow of a compressed gas must be controlled for practical and safe use.

Answers

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The Periodic Table and
Periodic Law Section 6.1
Development of the
Modern Periodic Table
1. octaves 2. eight 3. nine
4. accepted 5. Dmitri
Mendeleev 6. atomic
mass 7. elements 8.
atomic number 9. Henry
Moseley 10. protons 11.
periodic law 12.
properties 13. 14 ...

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answers

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Assessment Chemistry:

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Chapter 5 25 Electrons in

Atoms Reviewing

Vocabulary Match the

definition in Column A

with the term in Column

B. Column A Column B

1. The set of frequencies

of the electromagnetic

waves emitted by the

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atoms of an element 2.

The minimum amount of energy that can be lost or gained by ...

Answers

Chapter Assessment

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numbers used in
chemistry can be q... the
study of energy changes
(that is, energy produced
or absor... A system that
can exchange neither
energy nor matter with
its... Chemistry matter
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intensive property.

extensive properties

(examples) intensive

property (examples) a

property that depends on

the amount of matter in a

sample. a property that

depends on the type of

matter in a sample not....

mass, volume, length,

weight.

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moles of known

substance C H C H C H

O O O CO CO CO H O

H O H O number of

moles of unknown

substance O CO H O C

H CO H O C H O H O

C H O CO use mole

ratio 5 mol O 2 1 mol C 3

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H 8 3 mol CO 2 1 mol C

3 H 8 4 mol H 2 O 1 mol

C 3 H 8 1 mol C 3 H 8 5

mol O 2

Answers

MATH SKILLS

TRANSPARENCY

MASTER 16

A “ chemical change ”
is a change that produces
matter with a different
composition than the
original matter. Chemical
Change A change in

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which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component elements.

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