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Molarity. a unit describing CONCENTRATION of a solution.
solutions. homogeneous mixture of 2 or more pure substances. a
solution is composed of. solvent & solute. solvent. in greater quantity
than solute, typically a liquid but not always. solute.

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14. a 15. d 16. d 17. b 18. b...

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240 Chemistry: Matter and Change • Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

States of Matter

CHAPTER STUDY GUIDE FOR CONTENT MASTERY

Section 12.2 Stoichiometric Calculations In your textbook, read about mole-to-mole conversion. Read the following passage and then solve the problems. In the equation that follows each problem, write in the space provided the mole ratio that can be used to solve the problem.

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exercises for you to complete as you read the text. The study guide pages are divided into sections that match those in your text.

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