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~~SN1/SN2/E1/E2~~ Packet

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SN2, E1, \u0026

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E2 Reaction

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reaction

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~~Equations (part~~

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an element that

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is not equal on both sides of the equation.

Add a

coefficient (the large number to the left of the formula) in front of the formula with that element and adjust your counts. Continue adding

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coefficients
until the
numbers of
elements on each

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an element that
is not equal on
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Answer sides of
the equation.

Add a

coefficient (the
large number to
the left of the

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Answer

Pick an element that is not equal on both sides of the equation. Add a coefficient (the large number to the left of the formula) in front of the

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formula with
that element and
adjust your
counts. Continue
adding
coefficients
until the
numbers of
elements on each
side of the
arrow are equal.

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an element that
is not equal on
both sides of
the equation.

Add a
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large number to the left of the formula) in front of the formula with that element and adjust your counts. Continue adding coefficients until the numbers of elements on each side of the

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Answer

The study of the
quantitative
relationships
between the
amounts of
reactants used
and the amounts

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of products Packet

formed by a
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chemical
Answer is

called

stoichiometry.

2. Stoichiometry

is based on the

law of

conservation of

mass. 3. In any

chemical

reaction, the

mass of the

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products is less than the mass of the reactants.

4.

Stoichiometry

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1) What are some signs a chemical reaction has taken place? 2)

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Give an example of a chemical equation and label the

reactants and product. 3) What happens to the atom arrangement when a substance goes through a chemical reaction? 4)

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certain fires,
but hydrogen is
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