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AP Chemistry Lab #5 Analysis of

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Equation - Hydrogen Peroxide Calcium
Carbonate Decomposition Reaction

(Thermal) Rate of Reaction of Sodium
Thiosulfate and Hydrochloric Acid

Heating calcium carbonate - Suck Back

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decomposition of calcium carbonate

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Answers
iron(II) under acidic conditions

Decomposition of hydrogen peroxide The

Kinetic Analysis of the Decomposition of

Hydrogen Peroxide (Evaluation of Data)

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investigate the concentration of calcium
ions in hard water and consumer products
using gravimetric analysis. The purpose of

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of metals and a series of nonmetal halogens are studied to find their relative reactivities. The reactivity of the metals is determined by combining the metals with a complementary series of metal ions in solution. The reactivity of three ...

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Oxidation-reduction reactions form a
major class of chemical reactions. From
the reactions of oxygen with sugars, fats,

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~~Answers~~ and proteins that provide energy for life to the corrosion of metals, many important reactions involve the processes of oxidation and reduction. Page 5/25. Acces PDF Chemfax Lab 21 Answers FLI SCIENTIFIC ...

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After the calculations, I have evaluated the molar volume and density for trials one and two: 8.25 L/mol H₂ and 0.244 g/L H₂; 9.13 L/mol H₂ and 0.220 g/L H₂, respectively. Compared to the actual values, I was off for the molar volume by 63% and 59% for each trial, respectively. For the density, I was off by 29% and 26%

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Answers for each trial, respectively, a percentage lower than the molar volume ...

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