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## Chapter 2: Introduction to the Chemistry of Life ...

Chapter 2 The Chemistry of Life What are the basic building blocks of all matter? Atoms Describe the structure of an atom protons, electrons, and neutrons. The nucleus (center) of the atom contains

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the protons (positively charged) and the neutrons (no charge).

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What are the basic chemical principles that affect living things?

Biology Chapter 2- The Chemistry of Life

Biology: Chapter 2, The Chemistry of Life. Atom. Nucleus.  
Electron. Element. An atom is the smallest constituent unit of  
ordinary matter th.... The nucleus is the small, dense region  
consisting of protons a.... The electron is a subatomic particle,  
symbol  $e^-$  or  $e^{-}$ , with a....

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A B; atom: the basic unit of matter: nucleus: the center of the atom: electron: a negatively charged particle: element: a pure substance that consists entirely of one type of atom

## Quia - Chapter 2: The Chemistry of Life Vocabulary Review

Chemistry 1405 Chapter 2 1. chemistry. Chemistry is the study of matter, its properties, how and why substances combine or separate to form other substances, and how substances interact with energy. 2. matter. The term matter refers to anything that occupies space and has mass—in other words, the “ stuff ” that the universe is made of. 3. three different levels of matter – The three ...

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A compound is a substance formed by the chemical combination of

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two or more elements in definite proportions. (this means that  $H_2O$  is water, but  $H_3O$  is not). Compounds have different physical and chemical properties from the elements they are made of

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this picture? Is it just a mass of tangled ribbons? Look closely. It ' s  
actually a complex pattern of three-dimensional shapes. It  
represents the structure of a common chemical found inside living  
cells. The chemical is a protein called kinase.

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2. What relationship exists between the mass number of an element and isotopes of that element? 3. Explain the difference between ionic and covalent bonds. 4. Compare and contrast adhesion and cohesion. Below is a guide for Chapter 2 Test

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This chapter looks at atoms, bonds, pH and organic molecules. Good review of chemistry we see in microbiology.

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Chapter 2 The Chemistry of Life Reviewing Key Concepts Class Date Section Review 2-4 Completion On the lines provided, complete the following sentences. 1. Chemical reactions that energy often occur spontaneously. 2. During a chemical reaction, chemical

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bonds are 3. Biological catalysts, or enzymes, act by lowering the required for a reaction. 4.

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