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used in cannabinoid production, purification, and distillation. Through years of experience, we found manufacturers lacked a single, reliable source for acquiring chemicals used to make cannabis oil. We solve the problem by providing a variety of proven, hand-selected products for the industry.

~~Carbon Chemistry LTD - Homepage~~

The simplest organic chemicals, called hydrocarbons, contain only carbon and hydrogen atoms; the simplest hydrocarbon (called methane) contains a single carbon atom bonded to four hydrogen atoms: Methane -

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a carbon atom bonded to 4 hydrogen atoms.

~~Carbon Chemistry | Chemistry | Visionlearning~~

Carbon (from Latin: carbo "coal") is a chemical element with the symbol C and atomic number 6. It is nonmetallic and tetravalent—making four electrons available to form covalent chemical bonds. It belongs to group 14 of the periodic table. Carbon makes up only about 0.025 percent of Earth's crust.

~~Carbon - Wikipedia~~

Carbon, chemical element that forms more compounds than all the other elements

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combined. Carbon is widely distributed in coal and in the compounds that make up petroleum, natural gas, and plant and animal tissue. The carbon cycle is one of the most important of all biological processes.

~~carbon | Facts, Uses, & Properties |  
Britannica~~

The chemistry of carbon is dominated by three factors. 1. Carbon forms unusually strong C-C single bonds, C=C double bonds, and carbon-carbon triple bonds. 2. The electronegativity of carbon (EN = 2.55) is too small to allow carbon to form C<sup>4-</sup> ions with most metals and



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too large for carbon to form  $C^{4+}$  ions when it reacts with nonmetals. Carbon therefore forms covalent bonds with many other elements.

~~The Chemistry of Carbon — Purdue University~~  
Carbon Chemistry is an introduction to organic chemistry and biochemistry for ages 9-14. It is intended as a follow-up to "The Elements" by the same author, but can be used separately if the student has some basic knowledge of chemistry and the Periodic Table.

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~~Carbon Chemistry: McHenry, Ellen Johnston:  
9780982537725 ...~~

Description: This curriculum is designed to be a follow-up to "The Elements.". If you've finished that curriculum and want even more chemistry, here is a unit all about carbon-based chemistry. The style of the text (including the bits of whimsical humor) are the same as in "The Elements," but the student text is more substantial (83 pages) and therefore requires more reading.

~~Carbon Chemistry (Digital Download) — Ellen  
McHenry's ...~~

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One of the most important compounds of carbon is the charcoal, which is formed when carbon is heated in the absence in of air. It occurs in a number of allotropic forms. Allotropes are nothing but forms of an element with varying physical as well as chemical properties.

~~Physical And Chemical Properties Of Carbon | Chemistry ...~~

Carbon chemistry is so important that it has a whole branch of chemistry entirely devoted to it - organic chemistry. The number of compounds that contain carbon vastly exceeds

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all other compounds combined.

### ~~Carbon chemistry~~

Carbon chemistry Pure carbon can be found as diamond, graphite, fullerenes and graphene. Crude oil is made from hydrocarbons which can be processed to make a variety of fuels and other useful...

### ~~Diamond and graphite — Carbon chemistry — AQA Synergy ...~~

Indeed, carbon's compounds are so numerous, complex, and important that their study constitutes a specialized field of chemistry

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called organic chemistry, which derives its name from the fact that in the 19th century most of the then-known carbon compounds were considered to have originated in living organisms.

### ~~Carbon — Compounds | Britannica~~

Carbon is a common component of which of the following items? a. air; b. iron; c. people; d. water; Organic chemistry is broadly defined as: a. the chemistry of metallic compounds; b. the chemistry of substances produced by living organisms; c. the chemistry of the compounds that contain

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carbon and hydrogen; d. all of the choices;  
The simplest ...

~~Carbon Chemistry | Chemistry | Quiz |~~  
Visionlearning

English Language Learners Definition of carbon. : a chemical element that forms diamonds and coal and that is found in petroleum and in all living plants and animals. See the full definition for carbon in the English Language Learners Dictionary.

~~Carbon | Definition of Carbon by Merriam-~~  
Webster

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Carbon chemistry is the chemistry of life. It involves many different families of compounds and the reactions needed to interchange them. Understanding the properties of different compounds caused...

~~Carbon chemistry — Systematic carbon chemistry — Higher ...~~

chemical industry: Carbon black. The outstanding black pigment is the versatile product known as carbon black. Carbon black is one of the most important... Carbon black particles are usually spherical in shape and less regularly crystalline than graphite.

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~~Carbon black | chemistry | Britannica~~

Carbon nanotube chemistry involves chemical reactions, which are used to modify the properties of carbon nanotubes (CNTs). CNTs can be functionalized to attain desired properties that can be used in a wide variety of applications. The two main methods of CNT functionalization are covalent and non-covalent modifications.

~~Carbon nanotube chemistry - Wikipedia~~

Decarboxylation is a chemical reaction that removes a carboxyl group and releases carbon



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dioxide (CO<sub>2</sub>). Usually, decarboxylation refers to a reaction of carboxylic acids, removing a carbon atom from a carbon chain. The reverse process, which is the first chemical step in photosynthesis, is called carboxylation, the addition of CO<sub>2</sub> to a compound. Enzymes that catalyze decarboxylations are called ...

~~Decarboxylation — Wikipedia~~

This is the structure of carvone, with the chiral carbon noted by \*. I don't understand why this is a chiral carbon, to my understanding a chiral carbon is a carbon

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with 4 different chemical groups attached to it. In the molecule, the chiral carbon is surrounded by  $\text{H}$ ,  $\text{C}(\text{CH}_3)=\text{CH}_2$  and two  $\text{CH}_2$  groups.

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