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Precision. Precision is a measure of how close successive measurements are to each other. Precision is influenced by the scale, and when reporting a measurement, you report all certain values, and the the first uncertain one (which you "guesstimate"). This is illustrated in Figure 1B.2.2.

~~1B.2: Making Measurements: Experimental Error, Accuracy ...~~

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The random error will be smaller with a more accurate instrument (measurements are made in finer increments) and with more repeatability or reproducibility (precision). Consider a common laboratory experiment in which you must determine the percentage of acid in a sample of vinegar by observing the volume of sodium hydroxide solution required to neutralize a given volume of the vinegar.

~~Accuracy, Precision, and Error |~~

~~Introduction to Chemistry~~

There are two concepts we need to understand in experimental error, accuracy and precision. Accuracy is how close your

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value or measurement is to the correct (true) value, and precision is how close repeated measurements are to each other.

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