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Biology Chapter 2 Section 1 Study Guide. STUDY. PLAY. The basic unit of matter is called a(an) atom. Describe the nucleus of an atom. It is the center of the atom made up of protons and neutrons. Name the three subatomic particles, list their charge, and their location.

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Chapter 2 Biology Study Guide. atom's central core. negatively charged, surround the nucleus in a "cloud" ; orbit the nucleus Can only have 2 electron charges on shell. - orbit the nucleus of an atom in specific electron shells. - number of \_\_\_ in the outermost shell determines the chemical properties of an atom.

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2. Covalent Bonds Covalent bonds form through the process of sharing of electrons. Two atoms can fill their outer shells by sharing electrons. In fact, atoms give up little by sharing. For example, 2 Hydrogen atoms share their electron to have two electrons in their shell. Oxygen (1s<sup>2</sup>, 2s<sup>2</sup>, 2p<sup>4</sup>) shares two electrons to form O<sub>2</sub>. Methane

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