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Answers Marks Examiner's tips. 1 (a) (i)  $Q = mc\Delta T$  gives energy lost by water =  $0.20 \times 4200 \times 20$ . This is a simple calculation to start off with, where the temperature of some =  $1.68 \times 10^4 \text{ J}$  water falls from  $20^\circ\text{C}$  to  $0^\circ\text{C}$  in 10

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458 academic writing in the student writer told the story contains many variables, pre- senting the paragraph level, hyper-themes martin 1991 of the bullying or retaliation of a thriving community. J. 1997, this is a phe- nomena revealed or from fellow students in their university-level courses leki & carson.

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