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Refer to the photoelectron spectrum of neon shown below to answer the question. Which of the following statements best accounts for peak A being to the left of

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peaks B and C? answer choices

AP Chem PES | Chemistry Quiz - Quizizz

1. Mass of empty container 22.47 Initial mass of sample and container 25.29 Mass of sample and container after heating 24.43g Student 1 is assigned the task of determining the number of moles of water in one mole of $\text{CuSO}_4 \cdot n\text{H}_2\text{O}$ a) determine grams of anhydrous product b) calculate total number of moles of water lost when the sample was heated c) determine the formula of the hydrated compound ...

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Question: 143. 9. What is the work involved if a gas in a 2.0-liter container at 2.4 atmospheres pressure is allowed to expand against a pressure of 0.80 atmosphere? A. -3.2 L atm; B. -9.62 L atm; C. -4.8 L atm; D. +14.4 L atm; Correct Answer: A. Explanation: (A) Work can be determined from PV data. The equation is. work (w) = - P ? V.

AP Chemistry Question 143: Answer and Explanation_crackap.com

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Exam questions assess the course concepts and skills outlined in the course

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framework. For more information on exam weighting, download the AP Chemistry Course and Exam Description (CED).. Encourage your students to visit the AP Chemistry student page for exam information and exam practice.

AP Chemistry Exam - AP Central | College Board

AP Chemistry Question? 2.920 g of a Group II metal, X, reacts with an excess of chlorine to form 5.287 g of a compound with formula XCl_2 . What is metal X? 1
Answer

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Question 3 (continued) 3 33 23 3 2 3 3 1

mol CaCO 0.93 g CaCO 0.0093 mol

CaCO 100.1 g 1 mol Na CO ... AP

Chemistry Scoring Guidelines from the 2019 Exam Administration Author:

College Board Subject: AP Chemistry

Scoring Guidelines from the 2019 Exam

Administration Keywords:

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answer. $q = mc \Delta T = (110.0 \text{ g})(4.18$

$\text{J}/(\text{g}\cdot^\circ\text{C}))(35.6^\circ\text{C} - 15.0^\circ\text{C}) = 9,470 \text{ J} =$

9.47 kJ 1 point is earned for the correct

setup. 1 point is earned for the correct

answer with units. (ii) Determine the value

of H_{soln} for LiCl in kJ/mol rxn. 1 mol

LiCl 10.0 g LiCl 0.236 mol LiCl 42.39 g

LiCl 9.47 kJ 0.236 mol LiCl = 40.1 kJ/mol

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rxn soln

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