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and Thermo 11 1. Heat is  
added to a gas. A.  $Q$  for the  
gas is: B. Give one way that  
the gas could not change its  
temperature. C. \* True or  
false and why: "A gas's  
temperature must decrease

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when it releases heat." 2. 5 kg of an unknown substance requires 60kJ to vaporize completely. What is the latent heat of ...

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 $Q_H$  arrow is equal to the  $Q_C$   
and  $W$  arrows. So  $Q_H$  is  
the whole and the others are  
the parts.  $W$  is the useful  
work done by the engine.  $Q_C$   
is the heat that is not

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used, so is "wasted" and is the inefficiency of the engine. This engine is 50% efficient.

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2. In one complete cycle a

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heat engine absorbs  $450 \text{ J}$  of heat from a hot reservoir (a reserve of heat) and expels  $200 \text{ J}$  of heat to a cold reservoir (a reserve that stays cold). What is the efficiency of the engine? 3. A heat engine is  $45\%$

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efficient. It absorbs  $600\text{J}$  of heat each second from a hot reservoir. A. What percent is  $600\text{J}$ ? B.

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Dimensionally Stable - The  
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intensive heat the wood is exposed to during its production dries out deep into the core, causing all moisture and resin to be removed from the timber. Thermowood therefore doesn't react to changes in humidity

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as drastically as untreated wood (the risk of swelling, cracking and shrinkage is decreased), allowing it to retain its shape far better.

Thermowood - The Ultimate  
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Thermoregulation, also called Heat Regulation, the maintenance of an optimum temperature range by an organism. Cold-blooded animals (poikilotherms) pick up or lose heat by way of

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the environment, moving from one place to another as necessary. Warm-blooded animals (homoiotherms) have additional means by which they can heat and cool their bodies.

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